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# ISOPRENOID TRIETHER LIPIDS FROM CALDARIELLA

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Key Word Index—Caldariella; lipids; ether lipids; isoprenoids; tri-O-phytanylglycerol.

In the Caldariella group of extreme thermoacidophile bacteria [1], the major lipids [2,3] are macrocyclic sn-1,2-diethers of glycerol derived from a range of  $C_{40}$  diols with unique  $\omega,\omega'$ -biphytanyl structures. In the hope of shedding some light on their biosynthesis, we investigated minor lipid components, here shown to be triethers of glycerol with partly-or fully-saturated  $C_{20}$  (phytanyl) alcohols.

# Separation of minor lipids

From ca 350 g of lyophilized cells of the MT-3 and MT-4 strains of Caldariella acidophila, some 25 g of total lipid was obtained, of which ca 0.5 g (accumulated over a series of experiments) was directly soluble in hexane and separated accordingly. Chromatography of this fraction afforded the thianaphthenequinone described elsewhere [4] as major component, a complex mixture of hydrocarbons, and some 30 mg of glycerol triethers. These on TLC (Si gel; hexane-Et<sub>2</sub>O, 7:3) afforded two components,  $R_f$  0.5 and 0.4; the former predominated in the triether fraction from MT-4.

# Tri-O-phytanylglycerol (1)

The IR spectrum of the less polar triether showed only saturated alkyl and ether bands (v<sub>m</sub> 1110, 1375-1385, 1465, 2850-2950 cm<sup>-1</sup>) and the 1H NMR spectrum showed, per 11 CH<sub>2</sub>-O and CH-O protons ( $\delta$  3.40, br) some 72 saturated CH<sub>2</sub> and CH protons ( $\delta$ 1.22–1.60, br) and 45 CH<sub>3</sub> protons ( $\delta$  0.86, d, J = 9Hz). Satisfactory MS were not obtained. Treatment with BCl3 gave glycerol (TLC) and an alkyl chloride C20H41Cl (by MS); similarly HI cleavage gave C<sub>20</sub>H<sub>41</sub>l. The <sup>1</sup>H NMR spectra of these halides showed, per 2 CH<sub>2</sub>X protons ( $\delta$  3.15, t) 24 CH<sub>2</sub> and CH protons ( $\delta$  1.22-1.60) and 15 CH<sub>3</sub> protons ( $\delta$  0.86, d, J = 9 Hz) and their identification as phytanyl halides was confirmed by conversion into the acetate which was identical (MS, NMR, and GLC) with authentic phytanyl acetate. Thus structure (1) for the less polar triether is established.

#### The more polar component

The more polar triether component was very similar to (1) but contained some unsaturation (addition  $v_m$  1665 cm<sup>-1</sup>; NMR data below). Hydrogenation gave a product identical with (1) which was similarly converted into

- (1) (no unsaturation)
- (2) (unsaturation as broken lines, see text)

authenticated phytanyl acetate. The extent and location of unsaturation in this component (which is probably a mixture) were in part extablished by <sup>1</sup>H NMR methods. If we regard (1) biogenetically as a tri-O-geranylgeranylglycerol which has been completely reduced, then in a partly-reduced analogue there are 12 possible double bond sites and these are of three distinguishable kinds: (i) in the  $\alpha$ -residue, bearing allylic CH<sub>2</sub>-O; (ii) in the  $\beta$ - and  $\gamma$ -residues; (iii) in the  $\delta$ -residue, with terminal =CMe<sub>2</sub>.

The resolved signals, assignments, and approximate integrals (calibrated from the total of the  $C\underline{H}_2$ -O and CHO signals,  $\delta$  3.40 and 3.95; 11 protons) in the <sup>1</sup>H NMR spectrum of the more polar triether fraction are:  $\delta$  0.86 (21H, d, J = 9 Hz,  $C\underline{H}_3$ -CH sat); 1.10-1.40 (36H, br,  $C\underline{H}_2$  and  $C\underline{H}$ , sat); 1.57 (6H, s, one  $C\underline{H}_3$ -CMe= in  $\delta$ -residue); 1.66 (18H, s,  $C\underline{H}_3$ -C= in  $\alpha$ ,  $\beta$ ,  $\gamma$  and one  $C\underline{H}_3$ -CMe= in  $\delta$ -residue); 2.00 (18H, br,  $C\underline{H}_2$ -CH=); 3.40 (9H, br, sat  $C\underline{H}_2$ -O and  $C\underline{H}$ -O); 3.95 (2H,d, J = 8 Hz, =CH-CH $_2$ -O); 5.05 (5H, br, = $C\underline{H}$ -CH $_2$ ); 5.35 (1H, t, J = 8 Hz, = $C\underline{H}$ -CH $_2$ O). The data are satisfied if in the more polar triethers there are (approximately)  $\delta$  double

bonds, viz. one out of the 3 possible in  $\alpha$ -residues, three out of the 6 possible in  $\beta$ - and  $\gamma$ -residues, and two out of the 3 possible in  $\delta$ -residues (cf. structure 2).

This at least suggests that reduction of the postulated precursor tri-O-geranylgeranylglycerol occurs progressively from the oxygenated end of the isoprenoid chains. Unfortunately the structures of these very minor triethers do very little to clarify the biogenesis of the  $\omega,\omega'$ -bi-phytanyl skeleton in the major cyclic diether lipids of Caldariella.

# **EXPERIMENTAL**

Culture methods and lipid extraction procedures are described elsewhere [1,3]. Direct extraction of the total lipid (25 g) with hexane gave a soluble fraction (540 mg) which was subjected to chromatography on Merck Kieselgel (70–230 mesh) in hexane with increasing proportions of  $Et_2O$ . The triethers eluted with 20%  $Et_2O$  and were further separated into (1) and (2) by TLC (Merck Kieselgel 60-F254) in hexane-ether (7:3) ( $R_f$  0.5 and 0.4 respectively).

Hydrogenation of (2). The triether (10 mg) was treated with H<sub>2</sub> on Pd-C (5%; 15 mg) in MeOH (3 ml) for 24 hr; chromatography of the product, as above, gave (1) (NMR, IR) which was subsequently converted as for the natural saturated triether.

Ether cleavage etc. Treatment of (1) (15 mg) with 57% HI (1 ml; 24 h reflux) and work-up (extraction into hexane, washing with H<sub>2</sub>O, aq. K<sub>2</sub>CO<sub>3</sub>, aq. NA<sub>2</sub>S<sub>2</sub>O<sub>3</sub>, and H<sub>2</sub>O, partition with hexane and 90% aq. MeOH, and TLC in hexane) gave

phytanyl iodide, C<sub>20</sub>H<sub>41</sub>l (MS and NMR). The product (10 mg) was refluxed with AgOAc (20 mg) in AcOH (2 ml) for 24 hr. The reaction mixture was taken up in Et<sub>2</sub>O, centrifuged, washed (H<sub>2</sub>O, aq. NaCl, aq. NaHCO<sub>3</sub>, aq. Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub>, H<sub>2</sub>O) and purified by TLC in hexane-ether (17:3). The product  $R_f$ 0.75, was identified as phytanyl acetate by comparison with authentic material [MS, NMR, and GLC (on a 2m × 3mm glass column packed with 1% OV-1 on Gas-Chrom Q 100/120with 40 ml/min N<sub>2</sub> at 200°; retention time 6.24 min)]. Alternatively, the triether (10 mg) was treated with BCl<sub>3</sub> (1 ml with 1 ml CHCl<sub>3</sub>; 14 hr at R.T.). Excess reagent was removed at reduced pressure and the alkyl chloride taken up in hexane for purification by TLC as above (phytanyl chloride, C<sub>20</sub>H<sub>41</sub>Cl by MS and NMR). The undissolved residue was taken up in methanol and identified as glycerol by TLC (in CHCl<sub>3</sub>-MeOH, 4:1 on Si gel,  $R_f$  0.40, visualised with Ag-NH<sub>3</sub>).

Spectra. All IR spectra were measured in CCl<sub>4</sub>; NMR spectra at 100 MHz in CCl<sub>4</sub> with TMS standard; MS on the AEI MS-30.

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# 6\$,22-DIHYDROXYHOPANE, A NEW TRITERPENE FROM THE FERN CHEILANTHES MARANTAE

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Plant. Cheilanthes marantae L. local name doradilla acanalada. Specimen no. 322 deposited in the Herbarium of the Botany Department, University of La Laguna. Source. Monte de las Mercedes, Tenerife, Canary Islands. Collected in June. Previous work. None.

Present work. In addition to saccharose, diplopterol, fernenol, sitosterol and sitosterol-o- $\beta$ -D-glucoside, the new natural triperpene  $6\beta$ ,22-dihydroxyhopane (1a), characterized by its IR, PMR and MS spectra, was isolated from the stems and leaves of Cheilanthes marantae. The hopane framework of (1a) was deduced from its MS which showed the same fragmentation pattern as that of zeorin (1d) and related compounds [1]. The large downfield shifts of the methyl groups at C-4 $\beta$ , C-8 $\beta$  and C-10 $\beta$  observed in the PMR spectrum (ca 0.35 ppm) compared with 22-hydroxyhopane are compatible only with the presence of a C-6 $\beta$  hydroxyl group [2]. These assumptions were chemically confirmed by partial synthesis of (1a) from zeorin.

1a R\*H; β-OH R'=OH
1b R\*H; β-OAc R'=OAc
1c R\*O R'=OH
1d R\*H; α-OH R'=OH

### **EXPERIMENTAL**

Mp's determined on a Koffler hot-stage apparatus, are uncorr. Optical rotations were measured in CHCl<sub>3</sub> and PMR spectra in CDCl<sub>3</sub> with TMS as internal reference.